

Mark schemes

1	(a) Alpha – two protons and two neutrons	1
	Beta – electron from the nucleus	1
	Gamma – electromagnetic radiation	1
	(b) Gamma	
	Beta	
	Alpha	
	<i>allow 1 mark for 1 or 2 correct</i>	2
	(c) any two from:	
	<ul style="list-style-type: none"> • (radioactive) source not pointed at students • (radioactive) source outside the box for minimum time necessary • safety glasses or eye protection or do not look at source • gloves • (radioactive) source held away from body • (radioactive) source held with tongs / forceps 	
	<i>accept any other sensible and practical suggestion</i>	2
	(d) half-life = 80 s	1
	counts / s after 200 s = 71	
	<i>accept an answer of 70</i>	1
	(e) very small amount of radiation emitted	
	<i>accept similar / same level as background radiation</i>	1
		[10]

2**Level 3 (5–6 marks):**

A detailed and coherent explanation is provided. The student gives examples that argue a strong case and demonstrate deep knowledge. The student makes logical links between clearly identified, relevant points.

Level 2 (3–4 marks):

An attempt to link the description of the experiment and the results with differences between the two models. The student gives examples of where the plum pudding model does not explain observations. The logic used may not be clear.

Level 1 (1–2 marks):

Simple statements are made that the nuclear model is a better model. The response may fail to make logical links between the points raised.

0 marks:

No relevant content.

Indicative content

- alpha particle scattering experiment
- alpha particles directed at gold foil
- most alpha particles pass straight through
- (so) most of atom is empty space
- a few alpha particles deflected through large angles
- (so) mass is concentrated at centre of atom
- (and) nucleus is (positively) charged
- plum pudding model has mass spread throughout atom
- plum pudding model has charge spread throughout atom

[6]**3**

- (a) cannot predict which dice / atom will 'decay'

accept answers given in terms of 'roll a 6'

1

cannot predict when a dice / atom will 'decay'

1

- (b) 3.6 to 3.7 (rolls)

allow 1 mark for attempt to read graph when number of dice = 50

2

- (c) 90

1

- (d) uranium

1

- (e) beta

1

proton number has gone up (as neutron decays to proton and e^-)

1

(f) prevents contamination

or

prevents transfer of radioactive material to teacher's hands

1

which would cause damage / irradiation over a longer time period.

1

[10]

4

(a) Nucleus splitting into two fragments and releasing two or three neutrons

1

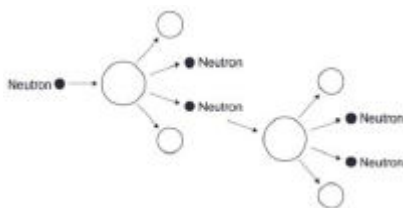
(at least one) fission neutron shown to be absorbed by additional large nucleus and causing fission

1

two or three additional neutrons released from fission reaction

1

This diagram would gain all 3 marks:



(b) lowering the control rods increases the number of neutrons absorbed

accept converse description

1

(so) energy released decreases

1

allow changing the position of the control rods affects the number of neutrons absorbed for 1 mark

(c) rate of increase between 240 and 276 (MW / min)

2

allow 1 mark for attempt to calculate gradient of line at 10 minutes

[7]

5

(a) neutrons and protons

1

(b) 0

1

(+)¹

1

(c) (i) total positive charge = total negative charge

accept protons and electrons have an equal opposite charge

1

(because) no of protons = no of electrons

1

(ii) ion

1

positive

1

- (d) Marks awarded for this answer will be determined by the quality of communication as well as the standard of the scientific response. Examiners should apply a best-fit approach to the marking.

0 marks

No relevant content

Level 1 (1 – 2 marks)

There is a basic description of at least **one** of the particles in terms of its characteristics.

Level 2 (3 – 4 marks)

There is a clear description of the characteristics of **both** particles

or

a full description of either alpha **or** beta particles in terms of their characteristics.

Level 3 (5 – 6 marks)

There is a clear and detailed description of **both** alpha and beta particles in terms of their characteristics.

examples of the physics points made in the response:

structure

- alpha particle consists of a helium nucleus
- alpha particle consists of 2 protons and 2 neutrons
- a beta particle is an electron
- a beta particle comes from the nucleus

penetration

- alpha particles are very poorly penetrating
- alpha particles can penetrate a few cm in air
- alpha particles are absorbed by skin
- alpha particles are absorbed by thin paper
- beta particles can penetrate several metres of air
- beta particles can pass through thin metal plate / foil
- beta particles can travel further than alpha particles in air
- beta particles can travel further than alpha particles in materials eg metals

deflection

- alpha particles and beta particles are deflected in opposite directions in an electric field
 - beta particles are deflected more than alpha particles
 - alpha particles have a greater charge than beta particles but beta particles have much less mass
- or**
- beta particles have a greater specific charge than alpha particles

6

- (a) (average) time taken for the amount / number of nuclei / atoms (of the isotope in a sample) to halve

or

time taken for the count rate (from a sample containing the isotope) to fall to half

accept (radio)activity for count rate

1

- (b) 60 ± 3 (days)

1

indication on graph how value was obtained

1

- (c) (i) cobalt(-60)

1

gamma not deflected by a magnetic field

or

gamma have no charge

dependent on first marking point

accept (only) emits gamma

gamma has no mass is insufficient

*do **not** accept any reference to half-life*

1

- (ii) strontium(-90)

1

any **two** from:

- *only* has beta
- alpha would be absorbed
- gamma unaffected
- *beta penetration / absorption depends on thickness of paper*
if thorium(-232) or radium(-226) given, max 2 marks can be awarded

2

- (iii) cobalt(-60)

1

shortest half-life

accept half-life is 5 years

dependent on first marking point

1

so activity / count rate will decrease quickest

1

- (iv) *americium(-241) / cobalt(-60) / radium(-226)*

1

gamma emitter

1

(only gamma) can penetrate lead (*of this box*)
do not allow lead fully absorbs gamma

1
[14]

7

(a) (i) neutron

1

(ii) neutron
proton

both required, either order

1

(iii) 2

1

number of protons

do not accept number of electrons

1

(b) (i) any **one** from:

- beta

- gamma

accept correct symbols

accept positron / neutrino / neutron

cosmic rays is insufficient

1

(ii) electrons

1

(iii) are highly ionising

1

(c) (i) mutate / destroy / kill / damage / change / ionise

Harm is insufficient

1

(ii) much smaller than

1

[9]

8

(a) J

reason only scores if J is chosen

1

(only) stars (about) the same / smaller size / mass as the Sun become black dwarfs

accept smaller than the Sun

accept it is the smallest

accept (only) small stars become black dwarfs

1

- (b) (i) become a supernova
or
 it will explode
ignore subsequent correct stages 1
- (ii) cannot take measurements needed
or
 do not have the technology
*do **not** accept cannot measure mass* 1
- (iii) advances in (measuring) techniques / technology / knowledge 1
- (c) any **five** from:
ignore any information up to the end of the main sequence
Apply the list rule if more than 5 points are made
- star expands (to become)
 - a red giant
red supergiant is incorrect
 - heavier elements are formed (by fusion)
elements heavier than iron are formed is incorrect
 - star shrinks (to become)
 - a white dwarf
supernova, neutron star, black hole are incorrect
 - star cools / fades
 - star stops emitting energy / radiation
star loses all energy is insufficient
- 5
[10]

- 9** (a) (same) number of protons
same atomic number is insufficient 1
- (b) (i) nuclei split
*do **not** accept atom for nuclei / nucleus* 1
- (ii) (nuclear) reactor 1
- (c) beta 1
- any **one** from:
- atomic / proton number increases (by 1)
accept atomic / proton number changes by 1
 - number of neutrons decreases / changes by 1
 - mass number does not change
(total) number of protons and neutrons does not change
 - a neutron becomes a proton 1
- (d) (average) time taken for number of nuclei to halve
or
(average) time taken for count-rate / activity to halve 1
- (e) (i) 6.2 (days)
Accept 6.2 to 6.3 inclusive
*allow **1** mark for correctly calculating number remaining as 20 000*
or
*allow **1** mark for number of*
80 000 plus correct use of the graph (gives an answer of 0.8 days) 2
- (ii) radiation causes ionisation
allow radiation can be ionising 1
- that may then harm / kill healthy cells
accept specific examples of harm, eg alter DNA / cause cancer 1
- (iii) benefit (of diagnosis / treatment) greater than risk (of radiation)
accept may be the only procedure available 1
- [11]**
- 10** (a) (i) nuclear reactor 1
- star 1

- (ii) nuclei are joined (not split)
accept converse in reference to nuclear fission
*do **not** accept atoms are joined*

1

- (b) (i) any **four** from:

- neutron
- (neutron) absorbed by U (nucleus)
ignore atom
*do **not** accept reacts*
*do **not** accept added to*
- forms a larger nucleus
- (this larger nucleus is) unstable
- (larger nucleus) splits into two (smaller) nuclei / into Ba and Kr
- releasing three neutrons and energy
accept fast-moving for energy

4

- (ii) 56 (Ba)

1

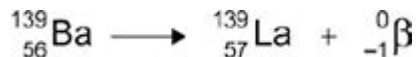
57 (La)

if proton number of Ba is incorrect allow 1 mark if that of La is 1 greater

1



accept e for β



scores 3 marks

1

[10]

11

- (a) (i) 18

1

- (ii) the count rate for the source

1

- (iii) the alpha radiation would not cover such a distance

1

- (iv) plots correct to within $\frac{1}{2}$ small square
allow 1 mark for 4 correct points plotted

2

correct curve through points as judged by eye

1

(v) two attempts at finding 'half-distance' using the table

20 to 10 cpm $d = 0.4 \text{ m}$

125 to 56 cpm $d = 0.2 \text{ m}$

31 to 14 cpm $d = 0.4 \text{ m}$

allow 1 mark for one attempted comparison

2

obeyed or not obeyed

dependent on previous two marks

1

(b) (i) there is no effect on the count rate in experiment 1 because the field is parallel
or beta particles are not deflected **or** there is no force

1

count rate is reduced in experiment 2 because field is perpendicular **or** beta particles are deflected **or** there is a force

1

(ii) only background radiation (as beta do not travel as far)

1

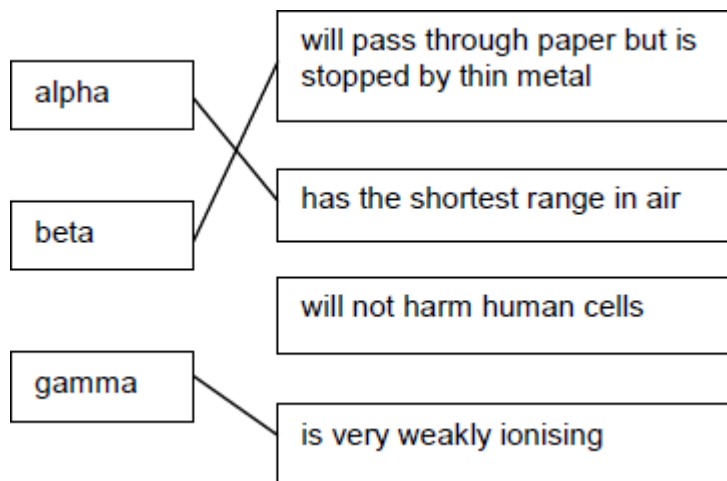
slightly different values show the random nature of radioactive decay

1

[13]

12

(a) 3 lines correct



allow 1 mark for each correct line

if more than one line is drawn from any type of radiation box then all of those lines are wrong

3

(b) Gamma radiation will pass through the body

1

(c) half

1

(d) protons

1

[6]

13

(a) uranium-235

accept any correct indication

1

(b) splits / breaks (into two smaller parts)

nucleus is separated is insufficient

*do **not** accept atom splits – on its own*

1

and (two / three) neutrons

1

(c) steam

correct order only

1

turbine

1

generator

1

[6]

14

(a) (i) plutonium (239)

accept Pu / Thorium / MOX (mixed oxide)

*do **not** accept uranium-238 **or** hydrogen*

1

(ii) (energy) used to heat water and

1

produce (high pressure) steam

1

the steam drives a turbine (which turns a generator)

1

(b) Neutron(s) shown 'hitting' other U-235 nuclei

one uranium nucleus is sufficient

1

U-235 nuclei (splitting) producing 2 or more neutrons

1

(c) any **two** from:

- neutrons are absorbed (by boron / control rods)
- there are fewer neutrons
- chain reaction slows down / stops
accept fewer reactions occur

2

[8]**15**

(a) 78

1

(b) atomic

1

(c) (i) 131

correct order only

1

54

1

(ii) 32 (days)

allow 1 mark for showing 4 half-lives provided no subsequent step

2

(iii) limits amount of iodine-131 / radioactive iodine that can be absorbed

accept increases level of non-radioactive iodine in thyroid

*do **not** accept cancels out iodine-131*

1

so reducing risk of cancer (of the thyroid)

accept stops risk of cancer (of the thyroid)

1

[8]**16**

(a) cosmic rays

1

radon gas

1

(b) (i) Radioactive decay is a random process

1

(ii) 19

1

(iii) 140

accept 159 – their (b)(i) correctly calculated

1

(iv) gamma

1

the count stayed the same

1

or

gamma does not have a charge

accept gamma is an electromagnetic wave

(so) gamma is not deflected / affected by the magnetic field

accept magnet for magnetic field

*do **not** accept is not attracted to the magnet*

last two marks may be scored for an answer in terms of why it cannot be alpha or beta

only answer simply in terms of general properties of gamma are insufficient

1

(c) lead absorbs (some of the) radiation

accept radiation cannot pass through (the lead)

or

less radiation emitted into the (storage) room

1

(d) Should radioactive waste be dumped in the oceans

1

[10]

17

(a) neutron discovered

1

(b) neutron

all 3 in correct order

electron

allow 1 mark for 1 correct

proton

2

[3]

18

(a) forces (within the star) are balanced

if specific forces are mentioned they must be appropriate

1

(b) (i) bigger the mass (of the star) the shorter the 'main sequence' period

accept bigger the star the shorter the time

1

(ii) any **one** from:

- insufficient evidence
- do not know (exact) amount of hydrogen in star
accept do not know (exact) mass of star
- time too long (to measure directly)
- may be other factors (not yet known) that determine length of 'main sequence' period
- values are based on theory / calculation

1

(iii) faster than

1

larger stars have a shorter 'main sequence' period so they must have the faster (rate of) nuclear fusion

there must be a link between shorter 'main sequence' and nuclear fusion, this may be implied from the first marking point

1

the end of 'main sequence' happens as the hydrogen in (the core of) a star is used up

or

(since) they use up hydrogen at a faster (rate)

accept more massive stars (are brighter so) release energy faster

1

- (c) Marks awarded for this answer will be determined by the Quality of Written Communication (QWC) as well as the standard of the scientific response. Examiners should also refer to the information in the [Marking guidance](#), and apply a 'best-fit' approach to the marking.

0 marks

No relevant content.

Level 1 (1-2 marks)

There is a basic description of what happens to a star much larger than the Sun after the 'main sequence' period.

OR

Two stages are correctly named and are in the correct sequence.

Level 2 (3-4 marks)

There is a clear description of what happens to a star much larger than the Sun after the 'main sequence' period.

AND

At least two stages are correctly named and are in the correct sequence.

Level 3 (5-6 marks)

There is a detailed description of what happens to a star much larger than the Sun after the 'main sequence' period.

AND

At least three stages are named, in the correct sequence. There are no additional incorrect stages given.

Examples of the points made in the response:

extra information

- (the core of the) star runs out of hydrogen
- (the star) expands (to form)
- (the star) cools (to form)
 - *the core shrinks*
 - *helium starts to fuse to form other elements*
- a red supergiant
 - accept super red giant*
 - do **not** accept red giant*
 - (outer layers) explode
 - *fusion of lighter elements to form heavier elements (up to iron)*
- as a supernova
 - elements heavier than iron are formed
 - accept heaviest elements are formed*
 - core shrinks
- becoming a neutron star

- if mass large enough (core collapses)
- (to form) a black hole
if a correct description and sequence for a star the same size as the Sun and much bigger than the Sun given without clearly indicating which is which is limited to Level 2

6
[12]

19

(a) (i) any **one** from:

- nuclear power (stations)
accept nuclear waste
accept coal power stations
- nuclear weapons (testing)
accept nuclear bombs / fallout
- nuclear accidents
accept named accident, eg Chernobyl or Fukushima
accept named medical procedure which involves a radioactive source
accept radiotherapy
accept X-rays
accept specific industrial examples that involve a radioactive source
nuclear activity / radiation is insufficient
smoke detectors is insufficient

1

(ii) (radioactive decay) is a random process

accept an answer in terms of background / radiation varies (from one point in time to another)

1

(b) any **one** from:

- (maybe) other factors involved
accept a named 'sensible' factor, eg smoking
- evidence may not be valid
accept not enough data
- may not have (a complete) understanding of the process (involved)

1

(c) (i) 2

1

2

1

(ii) 218

correct order only

1

84

1

(d) 3.8 (days)

*allow 1 mark for showing correct method using the graph provided
no subsequent steps**correct answers obtained using numbers other than 800 and 400
gain 2 marks provided the method is shown*

2

[9]**20**

(a) (both graphs show an initial) increase in count rate

accept both show an increase

1

(b) only the right kidney is working correctly

1

any **two** from:*if incorrect box chosen maximum of 1 mark can be awarded
reference to named kidney can be inferred from the tick box*

- count-rate / level / line for right kidney decreases (rapidly)
it decreases is insufficient
- count-rate / level / line for left kidney does not change
it does not change is insufficient
- radiation is being passed out into urine – if referring to right kidney
- radiation is not being passed out – if referring to the left kidney
- left kidney does not initially absorb as much technetium-99

2

[4]**21**

(a) protons, electrons

both required, either order

1

neutrons

1

electron, nucleus

both required, this order

1

(b) 2.7 (days)

allow 1 mark for showing correct use of the graph

2

- (c) put source into water at **one** point on bank
accept the idea of testing different parts of the river bank at different times

1

see if radiation is detected in polluted area
accept idea of tracing

or

put source into water at three points on bank (1)
 see if radiation is detected downstream of factory **or** farmland **or** sewage treatment works (1)

1

[7]

22

- (a) proton
all 3 in correct order

electron
*allow 1 mark for 1 correct do **not***

neutron
accept letters p, e, n

2

- (b) 9
reason only scores if 9 is chosen

1

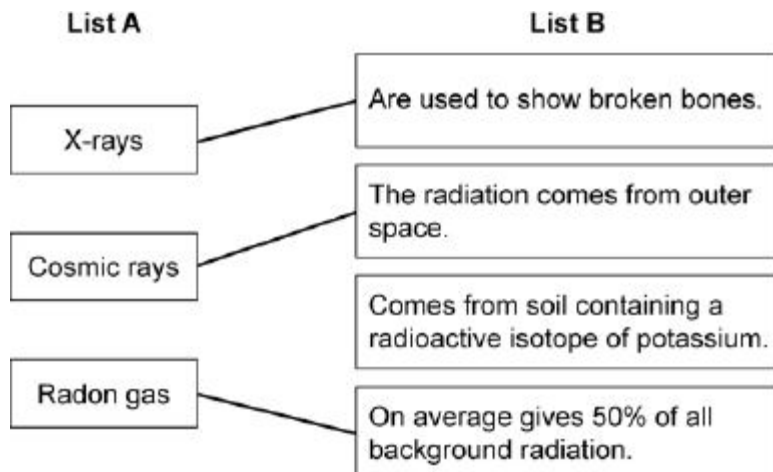
number of neutrons and protons

1

[4]

23

- (a) 1 mark for each correct line



*if more than 1 line has been drawn from a box in **List A** then all those lines are marked incorrect*

3

(b) higher in village B

1

by 6 units

allow 1 mark for correctly obtaining a height difference of 180 (m) / 4 times higher – this refers to height not radiation levels

accept for 3 marks in village A it is 2 units (extra) and in village B it is 8 units (extra) allow 1 mark for a correct radiation calculation based on incorrect height readings

2

[6]

24

(a) (i) 2.5

1

(ii) The radiation dose from natural sources is much greater than from artificial sources

1

(b) (i) any **one** from:

- different concentrations in different rooms
- to average out daily fluctuations
accept to find an average
accept to make the result (more) reliable / valid
do not accept to make more accurate on its own

1

(ii) average level (much) higher (in **C** and **D**)

accept converse

1

some homes have very high level (in **C** and **D**)

*accept maximum level in **A** and **B** is low*

1

or

maximum level in some homes (in **C** and **D**) is very high

*accept higher radiation levels (in **C** and **D**) for 1 mark*

[5]

25

any **two** pairs from:

to gain credit it must be clear which model is being described
*do **not** accept simple descriptions of the diagram without comparison*

- nuclear model mass is concentrated at the centre / nucleus (1)
*accept the nuclear model has a nucleus / the plum pudding model does not have a nucleus for **1** mark*

plum pudding model mass is evenly distributed (1)

- nuclear model positive charge occupies only a small part of the atom (1)

plum pudding model positive charge spread throughout the atom (1)

- nuclear model electrons orbit some distance from the centre (1)
accept electrons in shells / orbits provided a valid comparison is made with the plum pudding model

plum pudding electrons embedded in the (mass) of positive (charge) (1)
*do **not** accept electrons at edge of plum pudding*

- nuclear model the atom mainly empty space (1)

plum pudding model is a 'solid' mass (1)

[4]

26

(a) fusion

*do **not** credit any response which looks like 'fission'*

1

of hydrogen / H (atoms)

credit only if 1st mark point scores

1

(b) fusion of other / lighter atoms / elements

reference to big bang nullifies both marks

1

during supernova / explosion of star(s)

1

(c) the (available) evidence: supports this idea

or

does not contradict this idea

or

can be extrapolated to this idea

or

(electromagnetic) spectrum from other stars is similar to sun

1

[5]

- 27** (a) *answers must be in terms of nuclear fuels*
- concentrated source of energy
idea of a small mass of fuel able to generate a lot of electricity 1
- that is able to generate continuously
accept it is reliable
or can control / increase / decrease electricity generation
idea of available all of the time / not dependent on the weather
ignore reference to pollutant gases 1
- the energy from (nuclear) fission 1
- is used to heat water to steam to turn turbine linked to a generator 1
- (b) carbon dioxide is not released (into the atmosphere) 1
- but is (caught and) stored (in huge natural containers) 1
- [6]

- 28** (a) nucleus
- do **not** accept core / centre / middle* 1
- (b) radiation damages our cells
- accept radiation is dangerous / poisonous / harmful / toxic*
accept radiation can cause cancer / kills cells / change DNA / cause mutations / harm health
accept so precautions can be taken
accept so they know they may be exposed to / harmed by radiation
it refers to radiation (source)
to stop people being harmed is insufficient 1
- (c) **C** 1
- (d) gamma 1

gamma will pass through the lead

reason only scores if gamma chosen

or

alpha and beta will not pass through lead

accept correct symbols for alpha, beta and gamma

1

(e) (i) range of alpha too short

accept alpha would not reach detector

or

alpha absorbed whether box is full or empty

accept alpha (always) absorbed by box / card

accept alpha will not pass through the box / card

alphas cannot pass through objects / solids is insufficient

alpha not strong enough is insufficient

1

(ii) **M**

*reason only scores if **M** chosen*

1

less radiation / beta (particles) absorbed

accept more radiation / beta particles pass through

or

more radiation absorbed by full boxes

accept reading is higher

1

[8]

29

(a) **B E G**

all 3 required and no other

any order

1

same number of / 88 protons (and different numbers of neutrons)

same number of electrons is insufficient

1

(b) (i) 222

1

86

1

(ii) 4800

allow 1 mark for obtaining 3 half-lives

2

(c) ethical

1

deceived / lied to (about safety of working conditions)
accept (women) not warned of the dangers
given no protection is insufficient

or

value own / scientists' lives more than women

or

did not treat women humanely

1

- (d) accept any sensible suggestion
 eg
 too many interests in continued use of radium

evidence may cause public unrest
*do **not** accept not enough evidence*

doctors not want to be blamed for illnesses (caused by radium)
accept doctors not wanting to be sued (for harm caused by using radium)

doctors thought (possible) benefits outweighed (possible) risks
*do **not** accept did not know radium could be harmful*
believe radium could treat illnesses is insufficient

1

[9]

30

- (a) (i) (nuclear) fission is the splitting of a (large atomic) nucleus
*do **not** accept particle/atom for nucleus*

1

(nuclear) fusion is the joining of (two atomic) nuclei (to form a larger one)
do not accept particles/atoms for nuclei

1

- (ii) energy
accept heat/radiation/nuclear energy
accept gamma (radiation)
do not accept neutrons/neutrinos

1

- (b) (i) uranium (-235)
accept U (-235)
ignore any numbers given with uranium
accept thorium
accept MOX (mixed oxide)
*do **not** accept hydrogen*

1

- (ii) (same) number of protons
accept (same) atomic number
accept (same) positive charge
ignore reference to number of electrons

1

[5]

31

- (a) has an equal amount of positive charge
accept pudding/it is positive
- (b) (experimental) results could not be explained using 'plum pudding' model
or
 (experimental) results did not support plum pudding model
accept (experimental) results disproved plum pudding model
- (c) (i) **A** – most of atom is empty space **or** most of atom concentrated at the centre
- B** – nucleus is positive (so repels alpha particles)
accept nucleus has the same charge as alpha
- C** – nucleus is very small
accept nucleus is positive if not scored for B
or
 nucleus is a concentrated mass
accept nucleus has a very concentrated charge
- (ii) (if predictions correct, this) supports the new model
answers should be in terms of the nuclear model
accept supports his/new/nuclear theory
accept proves for supports
accept shows predictions/ Rutherford was correct

1

1

1

1

1

1

[6]

32

- (a) (i) protostar
correct order only
- red giant
- black dwarf

1

1

1

- (ii) Alpha Centauri A
accept any correct indication, eg alpha, centauri, A
reason only scores if Alpha Centauri A is chosen

1

stars (about) same size as Sun form white / black dwarfs

or

very large stars form red super giants / supernova/black hole

it is the same size as the Sun is insufficient

same life cycle as the Sun is insufficient

1

- (b) Atomic nuclei inside the star join together

1

[6]

33

- (a) (i) on average, cosmic rays produce less background radiation than rocks.

1

having no X-rays reduces a person's radiation dose.

1

- (ii) 4

allow 1 mark for 350 / 4

allow 1 mark for an answer 3.5

2

- (b) (i) (risk) increases

1

- (ii) C

reason only scores if C chosen

1

shows a lower risk for low doses (than for zero exposure)

accept risk reduces when you go from low to moderate (doses)

1

- (c) (i) *no mark for YES or NO, marks are for the explanation*

YES

fewer mice exposed first to a low dose

1

get cancer (than those only exposed to a high dose)
only scores if first marking point scores

NO

the results are for mice (1)

and may not be applicable to people (1)

1

(ii) ethical

1

[10]

34

(a) (i) 1.25 (mSv)

1

(ii) any **two** from:

- (frequent) flying
accept stated occupation that involves flying
- living at altitude
- living in areas with high radon concentrations
accept a specific area, eg Cornwall
- living in a building made from granite (blocks)
- having more than the average number of X-rays
 or
 having a CT scan
accept more medical treatments
- working in a nuclear power station
*accept any suggestion that could reasonably increase the level from
 a specific source*

2

(b) (i) to be able to see the effect of exposure (to radon gas)
or
 as a control

accept to compare (the effect of) exposure (with no exposure)

1

(ii) increased levels of exposure increases the risk (of developing cancer)
accept exposure (to radon gas) increases the risk

1

smoking increases the (harmful) effect of radon

answers that simply reproduce statistics are insufficient

1

(c) LNT model – risk increases with increasing radiation (dose) level

accept in (direct) proportion

accept low doses increase the risk

1

Radiation hormesis - low radiation (dose) levels reduce the risk

1

(d) two valid points made – examples:

- animals have no choice and so should not be used
 - should not make animals suffer
 - better to experiment on animals than humans
 - experiments lead to a better understanding / new knowledge
 - experiments may lead to health improvement / cures for humans
- results for animals may not apply to humans is insufficient*

2

[10]

35

(a) a protostar is at a lower temperature

or

a protostar does not emit radiation /energy

1

as (nuclear) fusion reactions have not started

accept heat or light for energy

1

(b) by (nuclear) fusion

accept nuclei fuse (together)

nuclear fusion and fission negates this mark

1

of hydrogen to helium

1

elements heavier than iron are formed in a supernova

accept a specific example e.g. heavier elements such as gold are formed in a supernova

accept heavier elements (up to iron) formed in red giant/red super giant

reference to burning (hydrogen) negates the first 2 marks

1

[5]**36**

(a) (i) 200 to 50

accept either order

1

(ii) 5.3

accept values between 5.2 and 5.4 inclusive

1

(iii) 5.3

accept values between 5.2 and 5.4 inclusive

or

their (a)(ii)

1

(b) (i) Make the conveyor belt move more slowly

1

(ii) lead

1

(c) Exposure increased the content of some types of vitamin.

1

[6]**37**

(a) cobalt-(60)

1

gamma (radiation) will pass through food / packaging

this can score if technetium chosen

1

long half-life so level of radiation (fairly) constant for (a number) of years

this can score if strontium / caesium is chosen

accept long half-life so source does not need frequent replacement

accept answers in terms of why alpha and beta cannot be used

gamma kills bacteria is insufficient

1

- (b) (i) people may link the use of radiation with illness / cancer
accept (they think) food becomes radioactive
accept (they think) it is harmful to them
'it' refers to irradiated food 1
- (ii) not biased / influenced (by government views) 1
- (iii) any **two** from:
 - data refers only to (cooked) chicken
 - data may not generalise to other foods
 - the content of some vitamins increases when food / chicken is irradiated
 - no vitamins are (completely) destroyed
 - (only) two vitamins decrease (but not significantly)
accept irradiated chicken / food contains a higher level of vitamins
marks are for the explanation only 2
- (iv) so can choose to eat / not eat that (particular) food
accept irradiated food may cause health problems
(for some people)
accept people may have ethical issues
(over eating irradiated food) 1
- (c) (i) electron
from nucleus / neutron
both parts required 1
- (ii) 90 years
allow 1 mark for showing 3 half-lives 2
- 38** (a) (i) half / 50 % 1
- (ii) Measure the radon gas level in more homes in this area 1
- (b) (i) 86 1

[11]

(ii) 222

1

[4]

39

(a) (i) (total) number of protons plus neutrons

*accept number of nucleons**accept amount for number**do not accept number of particles in the nucleus*

1

(ii) number of neutrons decreases by one

1

number of protons increases by one

accept for both marks a neutron changes into a proton

1

(b) (i) ${}_{81}^{208}\text{Th}$

1

correct order only

1

(ii) the number of protons determines the element

accept atomic number for number of protons

1

alpha and beta decay produce different changes to the number of protons

*there must be a comparison between alpha and beta which is more than a description of alpha and beta decay alone***or**

alpha and beta decay produce different atomic numbers

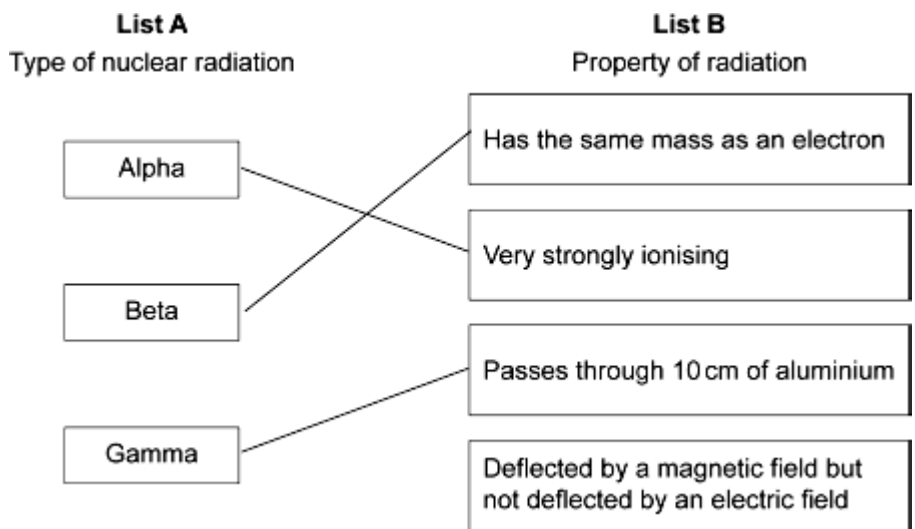
ignore correct reference to mass number

1

[7]

40

(a) 1 mark for each correct line



if more than 1 line is drawn from any box in List A, none of those lines gain any credit

3

(b) (i) (the detector) reading had gone down

'it' equals detector reading

accept the reading in the table is the smallest

accept 101 is (much) lower than other readings / a specific value eg 150

do not accept this answer if it indicates the readings are the thickness

1

more beta (particles / radiation) is being absorbed / stopped

accept radiation for beta particles / radiation

accept fewer particles being detected

1

(ii) six years

1

(iii) alpha would not penetrate the cardboard

accept the basic property – alpha (particles) cannot pass through paper / card

accept alpha (particles) are less penetrating (than beta)

range in air is neutral

1

[7]

41

(a) beta

1

alpha: would not pass through (the aluminium / foil)

1

gamma: no change in count rate when thickness changes

must be a connection between detection / count rate / passing through and change in thickness

1

(b) foil thickness increases then decreases (then back to normal / correct thickness)

a description of count rate changes is insufficient

1

gap between rollers decreases, then increases (then back to correct size)

or

pressure from rollers increases then decreases

accept tightness for pressure

answers may link change in thickness and gap width for full credit

ie:

foil thickness increases so gap between rollers decreases (1)

foil thickness decreases so gap between rollers increases (1)

1

(c) 56 (years)

accept any value between 55-57 inclusive

allow 1 mark for correct calculation of mass remaining as 1.5 (micrograms)

allow 1 mark for a mass of 4.5 micrograms plus correct use of graph with an answer of 12

maximum of 1 compensation mark can be awarded

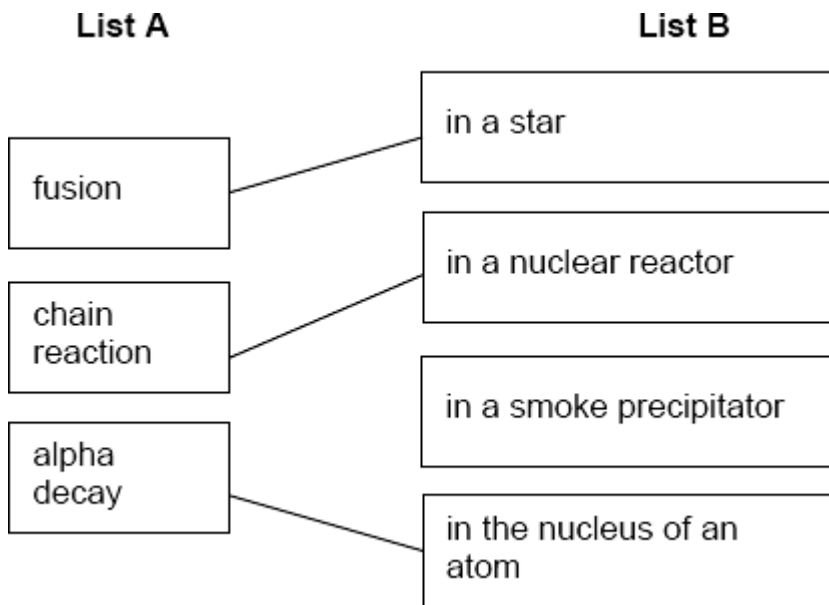
2

[7]

42 three lines correct

allow 1 mark for each correct line

if more than 1 line is drawn from a box in **List A**, mark each line incorrect



[3]

43 (a) electron(s)

1

(b) 3rd box ticked

The model cannot explain the results from a new experiment

1

(c) all three correct

Particle
Proton
Electron
Neutron

allow 1 mark for 1 correct

2

[4]

44

(a) (i) any **one** from:

- food / drink
- rocks / building materials
- cosmic rays / rays from space
accept correctly named example

1

(ii) any **one** from:

- nuclear power / coal power (stations)
accept nuclear waste
- nuclear accidents
accept named accident eg Chernobyl
- nuclear weapons testing
accept named medical procedure which involves a radioactive source
accept radiotherapy
nuclear activity / radiation is insufficient
*do **not** accept CT scans*

1

(iii) different number of / fewer protons

accept does not have 86 protons
accept only has 84 protons

or

different atomic number

*do **not** accept bottom number different*
reference to mass number negates this mark

1

(b) 168

accept 169 if clear, correct method is shown
*allow **1** mark for a correct dose ratio involving the spine*
eg 2:140 etc
***or** ratio of days to dose is 1.2*
***or** ratio of dose to days is 0.83*

2

(c) (i)

Group A	Group B
J M O	K L N

all correct

any order within each group

1

- (ii) similar (number) / same (number) / large (number)
accept the same specific number in each group eg three
reference to other factors such as age is neutral

1

- (iii) how many people in each group developed cancer
a clear comparison is required

1

- (iv) *there are no marks for **Yes** or **No** the mark is for the reason*

Yes

the benefit of having the scan is greater than the risk

or

the risk is (very) small (compared to the chance from natural causes)

accept the risk is much greater from natural causes

No

no additional risk is acceptable

1

[9]

45

- (a) gravitational attraction
accept 'gravity'
accept (nuclear) fusion

1

- (b) radiation 'pressure' and gravity / gravitational attraction
must be in correct context

1

are balanced / in equilibrium

accept are equal and opposite

*do **not** accept 'equal'*

or

there is sufficient / a lot of hydrogen / fuel

*do **not** accept constant supply of hydrogen*

to last a very long time / for (nuclear) fusion

this mark only scores if linked to the supply of hydrogen / fuel

reference to burning negates both marks

1

(c) (i) (conversion of) hydrogen to helium

accept (conversion of) lighter elements to heavier elements

1

by (nuclear) fusion

*note do **not** credit spelling of 'fusion' which could be 'fission'*

reference to burning negates both marks

1

(ii) massive supply / lots of hydrogen

1

(d) distributed throughout the Universe / space

*do **not** accept Solar System for Universe*

1

[7]

46

(a) (i) **L**

1

(ii) **M**

1

(b) To make a smoke detector work.

1

(c) **40**

no tolerance

1

[4]

47

(a) (i) number of protons are the same

accept atomic number / number of electrons for number of protons

1

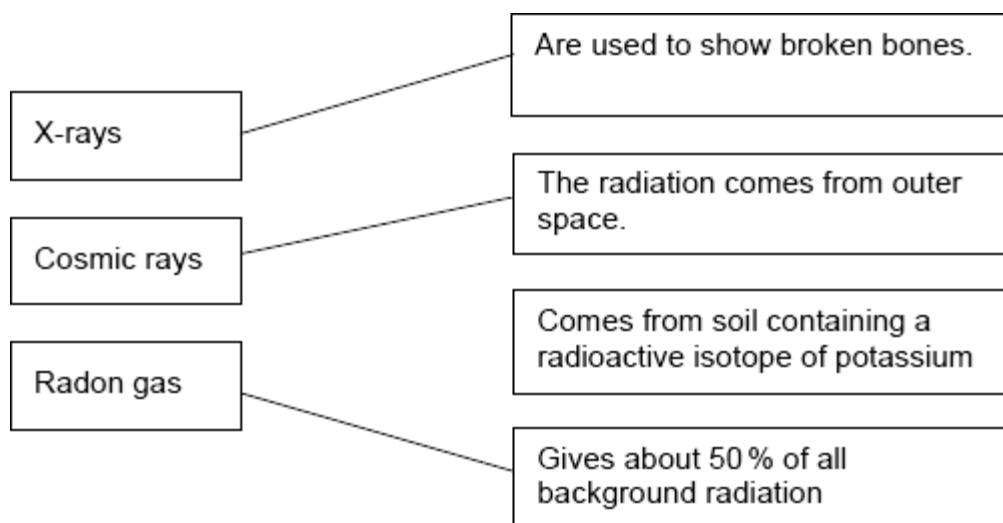
- number of neutrons are different
accept mass numbers are different – only if the first mark is awarded
 1
- (ii) an electron from the nucleus
both parts needed
 1
- (b) decays at the same rate as it is made
accept decays as fast as it is made
accept absorbed / used by plants (in CO₂) at same rate as it is being made
 1
- (c) (i) 3500
no tolerance
 1
- (ii) adjusted age correctly obtained from the graph
accept values between 3700–3800 inclusive
accept their (c)(i) used correctly to obtain an adjusted age from the graph
 1
- adjusted age +50
second mark can only be scored if first mark awarded
if no working shown an answer between 3750–3850 inclusive scores both marks
note: any line or mark made on the graph counts as working out
 1

[7]

48

1 mark for each correct line

*If more than 1 line has been drawn from a box in **List A** then all those lines are marked incorrect.*



[3]

49

- (a) proton
electron
neutron

*all 3 in correct order
allow 1 mark for 1 correct
do **not** accept letters p, e, n*

2

- (b) 4

reason only scores if 4 is chosen

1

number of protons

*accept number of electrons
accept there are 4 protons and 4 electrons
do **not** accept there are 4 protons and electrons*

1

- (c) The atom loses an electron.

1

[5]

- 50** (a) L
J
K
- all 3 in correct order
allow 1 mark for 1 correct*
- 2
- (b) number of electrons = number of protons
accept amount for number
- 1
- (c) neutrons
this answer only
- 1
- (d) loses / gains electron(s)
- 1
- [5]**

- 51** beta
- reason may score even if alpha or gamma given*
- 1

any **two** from:

- mass number does not change
or
total number of protons and neutrons does not change
 - atomic / proton number increases by 1
or
number of protons increases by 1
 - number of neutrons goes down by 1
- allow for 2 marks a neutron splits / changes into a proton and electron / beta*
- candidates that answer correctly in terms of why alpha **and** gamma are not possible, gain both marks*
- 2

[3]

- 52** (a) (forces due to) gravity and radiation pressure
- 1
- correct direction of forces
- 1

(forces) are balanced / equilibrium / equal

accept for 3 marks an answer in terms of

sufficient hydrogen (1)

to keep fusion reaction (1)

reference to burn / burning negates this mark

going at a continuous / steady rate (1)

if fuel is used instead of hydrogen maximum of 2 marks

1

(b) the Sun will remain stable (for several billion years)

1

based on evidence

accept a specific example of evidence

eg that the Sun has remained stable during the life of our planet / for 4.5 billion years

or

still contains more than 50 % hydrogen

or

by comparison with the lifecycle of (similar) stars

allow a refutation

eg not based on prejudice / whim / hearsay / folk law / historical or religious authority

1

[5]

53

(a) alpha particles **cannot** pass through...

*do **not** accept gamma particles...*

or

alpha particles can pass through a very thin sheet of **paper / card**

credit answers where correct amendments are made to boxed statement

1

(b) (i) horizontal and vertical line drawn at correct positions on the graph

accept a cross drawn at 4500 / 500 on the curve

or

two pairs of lines drawn, for example, at 600 and 300

accept a horizontal line drawn at 500 on its own

*do **not** accept vertical lines only*

1

(ii) 4500 million years

1

- (iii) half-life too long
do **not** accept simply its half-life is 4500 million years
1
- no (measurable) change in count rate
do **not** accept have not got the equipment
do **not** accept it's harmful (to children)
if neither of the above points scored, accept not enough time to measure it for **1** mark
1

[5]

54

- (a) (i) alpha (particle)
1
- (ii) (unstable) nucleus
accept (unstable) nuclei
do **not** accept middle
do **not** accept helium nucleus
1
- (iii) same number of protons
accept same number of electrons
accept same atomic / proton number
accept they both have 92 protons
same number of neutrons negates answer
1
- (b) (i) 4500 million years
do **not** accept 4500 years
1
- (ii) curve starting at 100 000 with a correct general shape
1
- passing through (4500, 50 000) and (9000, 25 000)
allow **1** mark for points plotted
or
line passing through (4500, 50 000) and (9000, 25 000)
1

[6]

55

- (a) (i) all correct
accept presented as a tally chart

Number of protons	3
Number of electrons	3
Number of neutrons	4

allow 1 mark for 1 correct

2

- (ii) 7
reason may score even if 7 not chosen

1

number of protons and neutrons
accept number of particles in the nucleus
accept number of nucleons
*do **not** accept number of electrons and neutrons*

1

- (b) an ion

1

- (c) (i) smaller than

1

- (ii) radon loses an alpha (particle)
or
radon loses an (alpha) particle
or
(mass of) polonium plus an alpha = (mass) radon
or
radon loses 2 protons and 2 neutrons (to become polonium)
accept radon has less protons and neutrons

1

[7]

56

- (a) (i) (two) nuclei (of light elements) join
accept hydrogen atoms for nuclei

1

forming a larger / heavier nucleus / one
accept comparative term equivalent to larger
accept forms a helium (nucleus / atom) this mark only scores if fusion is in terms of hydrogen atoms

1

- (ii) stars
accept a named star
e.g. the Sun
accept nebula
mention of planets negates answer

1

- (b) (i) any **one** from:
- (currently) only experimental
 - reaction does not last long enough
 - use more energy than they produce
allow difficult to control
*do **not** allow inefficient on its own*

1

- (ii) any **one** from:
- will give another source of energy
 - unlimited fuel supplies / energy
accept unlimited hydrogen
 - would not produce any radioactive waste
accept less radioactive waste
accept nuclear for radioactive
*do **not** accept toxic waste*
 - want to show that it can be done
accept any sensible suggestion
*do **not** accept answers only in terms of fossil fuels or carbon dioxide*

1

[5]**57**

- (a) (i) **K and L**
both answers required either order

1

- (ii) (1) same number of protons
accept same number of electrons
accept same atomic number

1

- (2) different numbers of neutrons

1

- (b) (i) 90 1
- (ii) 140 1
- (c) alpha (particle) 1
reason may score even if beta or gamma is chosen
- mass number goes down by 4
or
 number of protons and neutrons goes down by 4
or
 number of neutrons goes down by 2
*candidates that answer correctly in terms of why gamma
 and beta decay are not possible gain full credit* 1
- atomic / proton number goes down by 2
or
 number of protons goes down by 2
*accept an alpha particle consists of 2 neutrons and 2 protons for 1
 mark*
accept alpha equals ${}^4_2\text{He}$ or ${}^4_2\alpha$ for 1 mark
an alpha particle is a helium nucleus is insufficient for this mark 1
- [8]**

58

- (a) gravitational force(s) (1)
accept 'gravity'
- balanced by (force(s) due to) radiation pressure (1)
accept equal 2

(b) by (nuclear) fusion (1)

of hydrogen to helium (other light elements) (1)

allow 'low density' for light

accept hydrogen nuclei / atoms form helium

response must clearly link one element(s) producing others

fusion to produce helium (2)

heavy element / elements heavier than iron are only produced (by fusion) in a supernova (1)

allow dense for heavy

ignore any reference to elements undergoing radioactive decay (to form other elements)

3

[5]

59

(a) gamma will pass through the body

it refers to gamma

or

alpha will not pass through the body

answers must relate to the body

accept skin for body

1

gamma is only slightly ionising

accept gamma causes less damage to cells / tissue

or

alpha is heavily ionising

*do **not** accept gamma causes no damage to cells*

less harmful is insufficient

1

(b) (i) (both graphs show an initial) increase in count-rate

accept both show an increase

1

(ii) only the right kidney is working correctly

1

any **two** from:

*if incorrect box chosen maximum of 1 mark can be awarded
reference to named kidney can be inferred from the tick box*

- count-rate / level / line for right kidney decreases (rapidly)
it decreases is insufficient
- count-rate / level / line for left kidney does not change
it does not change is insufficient
- radiation is being passed out / into urine - if referring to right kidney
- radiation is not being passed out - if referring to the left kidney

2

- (c) (i) time taken for number of nuclei to halve
or
time taken for the count-rate to halve

1

- (ii) short half-life – the level of radiation (in the body) decreases rapidly
it refers to short life isotope

1

to a safe / very small level

or

a long half-life – the radiation remains in the body / for a long time

level of radiation remains high

answers in terms of damage eg cancer are insufficient

1

[9]

60

- (a) (i) 2.5

1

- (ii) The radiation dose from natural sources is much greater than from artificial sources.

1

- (b) (i) other factors may be involved

accept a specific suggestion

eg they may be exposed to other types of radiation

accept cannot be sure (in many cases) that the cause of death is radon (poisoning)

1

(ii) any **one** from:

- different concentrations in different rooms
- to average out daily fluctuations
accept to find an average
accept to make the result (more) reliable / valid
*do **not** accept to make more accurate on its own*

1

(iii) average level (much) higher (in **C** and **D**)

accept converse

1

some homes have very high level (in **C** and **D**)

*accept maximum level in **A** and **B** is low*

or

maximum level in some homes (in **C** and **D**) is very high

*accept higher radiation levels (in **C** and **D**) for **1** mark*

1

[6]