



## Mark schemes

<b>1</b>	(a) decrease <i>for 1 mark</i>	1	
	(b) (i) none would go through paper <i>for 1 mark</i>	1	
	(ii) all would go through paper <i>for 1 mark</i>	1	
	(iii) only some absorbed/amount absorbed depends on thickness of paper <i>for 1 mark each</i>	2	
	(c) $1 \rightarrow 1/2 \rightarrow 1/4 \rightarrow 1/8$ <i>for 1 mark</i>		
	3 half lives/ $3 \times 433$ <i>for 1 mark</i>		
	1299 years <i>gains 3 marks</i>	3	
			<b>[8]</b>
<b>2</b>	(a) (i) electron neutron proton nucleus <i>1 mark for each correct label</i>	4	
	(ii) H-1 has no neutrons H-3 has 2 neutrons <i>more neutrons gets 1 mark</i>	2	
	(iii) nucleus unstable	2	

- (b) lead/concrete  
lead/concrete needed to stop gamma rays

2

[10]

3

- (a) two from:  
internal/bodies  
thoron  
building materials  
soil  
food  
rocks  
radon  
gamma rays  
cosmic rays/outer space

*any 2 for 1 mark each*

2

- (b) only a very small amount of the background  
radioactivity comes from nuclear power owtte.

*accept any sensible response for 1 mark*

1

- (c) use G.M. tube/meter/counter or film  
'count' higher than or compare with background/normal/control or film is blacker

*for 1 mark each*

2

- (d) more neutrons/different number

*gains 1 mark*

**but** I-131 has 4 extra neutrons = 2

**or** I-131 has 78 neutrons I-127 has 74 (2)

*gains 2 marks*

2

- (e) (i) emits radiation  
ionises  
molecules  
in cells  
radiation damages cells/mutation/kills cells  
may cause cancer  
 $\beta$  /  $\gamma$  / radiation is penetrating  
half-life is long enough for damage to be caused

*any 4 for 1 mark each (4 from above)*

4

- (ii) e.g. replace I-127 in body/body cannot tell the difference/causes thyroid cancer/causes cancer (but not if already given in (i))  
*for 1 mark*
- (iii) **Either** No  
 half-life = 8 days many half lives have passed/attempts to calculate number of half-lives **or** explains meaning of half-life so very little left/become harmless  
*for 1 mark each*
- or** Yes half-life = 8 days such a large quantity was released although little left it is still harmful  
*for 1 mark each*

1

3

[15]

4

- (a) 1.  
 -1  
*for 1 mark each*
- (b) (i) 19p,  
 20n,  
 19e  
*all correct for 2 marks*  
*2 correct for 1 mark*
- (ii) K40 has an extra neutron/different number of neutrons/  
 it has more neutrons/21 neutrons  
*for 1 mark*  
NOT fewer neutrons
- (iii) radioactive/unstable nucleus/ nucleus disintegrates/  
 emits radiation/it has too many neutrons  
*for 1 mark*
- (iv) calcium/Ca  
*for 1 mark*
- (v) 1 (e) in outer shell/same number of electrons/outer electron  
 same distance from the nucleus  
*for 1 mark*

2

2

1

1

1

1

(c) (i) Geiger-Muller tube (photographic) film  
*for 1 mark*

1

(ii) cancer, leukaemia, radiation sickness etc.  
*for 1 mark*

1

**[10]****5**

(i) the nuclei  
of hydrogen/smaller atoms  
join to make helium/larger atoms  
*for 1 mark each*

3

(ii) the mass of the large nucleus (atom) is less than the mass of the smaller  
nuclei (atoms)  
*for 1 mark*

mass loss converted into energy or small mass loss given a large amount of energy  
*for 1 mark*

2

**[5]****6**

(a) 1, 0  
X, -1 (X = negligible / very small / (1/1840) to (1/2000), but not nothing  
*2 for 4 correct*  
*1 for 2/3 correct*

2

(b) has a nucleus which is positive charge  
negative charges (electrons) orbit nucleus  
*each for 1 mark*

3

**[5]**

7

- (a) two half lives

*gains 1 mark***but**

20 minutes

*gains 2 marks*

2

- (b) alphas will be stopped by skin / air
- or**
- do not penetrate betas and gammas can reach / damage organs / cells

*for 1 mark each*

2

**[4]**

8

- (a) nucleus positive charge / protons in nucleus electrons / negative charges orbit nucleus

*each for 1 mark*

3

- (b) (i) positive dough repels positive alpha particles
- or**
- 2 positive charges repel forces small

*each for 1 mark*

2

- (ii) large force needed + ves in plum pudding spread out –
- may appear in (i)*

positive charge must be concentrated / in nucleus

*(ignore references to electrons)**for 1 mark each*

3

- (c) 1, 0

X, -1 (X = negligible / very small / (1/1840) (1/2000), but not nothing)

*each row for 1 mark*

2

- (d) (i) 4

*for 1 mark*

1

- (ii) B and C have the same number of protons / atomic number but different number of neutrons / mass number

*each for 1 mark*

3

**[14]**





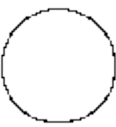
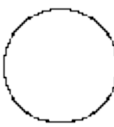
<b>9</b>	(a)	90	<i>for one mark</i>	1	
	(b)	(i)	neutron <i>for one mark</i>		1
		(ii)	nucleus <i>for one mark</i>		1
		(iii)	electron <i>for one mark</i>		1
	(c)	(i)	100 <i>for one mark</i>		1
		(ii)	157 <i>for one mark</i>		1
					<b>[6]</b>
<b>10</b>	(a)	(i)	B <i>for one mark</i>		2
		(ii)	has a different number of electrons (protons) <i>for one mark</i>		
	(b)	(i)	A and C <i>for one mark</i>		1
		(ii)	same number of protons / electrons, same nuclear charge different number of neutrons / nuclear masses different <i>for 1 mark each</i>		2
					<b>[5]</b>
<b>11</b>	(a)	(i)	beta and gamma ( <i>any order</i> ) <i>for one mark</i>		1

- (ii) gamma  
*for one mark* 1
- (b) (i) particles / atoms / molecules become charged / gain / lose electrons  
*for one mark* 1
- (ii) e.g. to kill cancer cells (*allow* any use of alpha, beta or gamma or X<sup>-</sup> radiation)  
*for one mark* 1
- (c) (i) time taken for no. of atoms / no. of nuclei / mass of U238 / activity to halve – **not** radioactivity  
**or**  
time taken for count rate to halve  
*for one mark* 1
- (ii) atoms with unstable nuclei which emit radiation  
(*not* definition of isotope but isotope which is radioactive gets 1 mark)  
*for 1 mark each* 2
- (d) (i) 1 / 4 *accept* 25% or 0.25  
*for one mark* 1
- (ii) 2 × half life or 2 × 4500 million years (independent of (i))  
gains 1 mark  
**but**  
9000 million years ecf only if answer to (i) is  $\frac{1}{2}, \frac{1}{8}, \frac{1}{16},$  etc.  
*gains 2 marks* 2
- [10]**

12

- (a) (i) B  
*for one mark* 1
- (ii) has 4 electrons / protons others only 3; B has a different no. of electrons / protons - *not* A and C have same no. of protons / electrons  
*for one mark* 1



- (b) (i) A and C  
for one mark 1
- (ii) same no. of protons / electrons different no. of neutrons  
**or**  
nuclei have the same charge but different mass  
for 1 mark each 2
- (c) (i)  3
- (ii) 
- (iii)   
for 1 mark each
- (d) 2p.2n  allow  but not   
(i.e. no mark if electrons shown)  
for one mark 1

[9]

13

- (a) the Sun is subject to two balancing forces / 2 forces in equilibrium  
the forces are: gravity making it contract **or** inward force due to gravity  
and a force due to temperature / heat / energy / radiation pressure making it  
expand **or** outward force due to temperature / heat / energy / radiation pressure  
for 1 mark each 3
- (b) Read all the answer first. Stop after 6 marks.
- hydrogen / fuel used up owtte the star will expand and become a red giant  
it will contract under gravity become a white dwarf  
it may explode and become a supernova throwing dust and gas into space  
leaving a dense neutron star / black hole  
(no mark for contradiction)  
any six for 1 mark each 6

[9]

**14** (i) 86

1

(ii) 222

1

[2]

**15** **Quality of written communication**

*The answer to this question requires ideas in good English in a sensible order with correct use of scientific terms. Quality of written communication should be considered in crediting points in the mark scheme*

Any **three** from

radon releases (alpha) radiation

*accept radon is radioactive*

1

(radon **or** radiation causes) harm **or** damage to body or cells

*accept cause cancer / mutations /  
radiation sickness*

1

idea that living near radiation over long period will lead to large 'dose' of radiation

radon (is a gas) that can be breathed in

1

[3]

**16** beta

1

alpha absorbed by paper

*allow beta and alpha  
second mark is linked to first*

1

**or** beta absorbed by aluminium allow beta can penetrate paper

**or** gamma would affect all of film

*i.e. cannot obtain second mark unless first mark is correct*

[2]

- 17** (a) radium  
*accept Ra* 1
- (b) different numbers of protons  
*accept one has 91 protons, one has 92*  
*or Pa has 91 protons, U has 92*  
*do not credit they have different atomic numbers*  
*reject different numbers of protons and neutrons* 1
- (c) alpha 1
- (d) neutron changes into proton  
*accept electron lost / beta radiation*  
*accept singular or plural answers* 1
- [4]**

- 18** any **three** from  
*max 2 if stages but no explanation*
- the star (Sun) expands because  
(inward) gravitational forces no longer balance (outward) force  
*accept the star collapses rapidly causing the core temperature to increase and the star to expand*  
*accept it expands because the forces are unbalanced*
  - to become a red giant
  - when the fusion stops it contracts / cools  
*accept (when hydrogen is used up) it collapses under gravity*  
*accept when fusion stops it contracts and explodes*
  - to become a white dwarf  
*accept to become a supernova / pulsar / neutron star / black hole*  
*(only if red giant has exploded)*

**[3]**

- 19**
- (a) (i) any **two** from  
 (matter from) exploded star / supernova  
 matter so dense / gravity so strong  
 that electromagnetic radiation / light cannot escape from it  
 2
- (ii) X-rays emitted  
 1  
 when gases or matter released from nearby stars spiral into it  
 1
- (b) fusion (of nuclei)  
 1  
 of lighter elements / hydrogen helium  
 1
- [6]**

- 20**
- (a) (i) cannot penetrate aluminium  
*allow can only pass through air / paper too weak is neutral*  
 1
- (ii) gamma rays not affected (by aluminium)  
*allow all / most (gamma rays) to pass through*  
*too strong is neutral*  
*danger is neutral*  
 1
- (b) (i) (nuclei) unstable  
 1
- (ii) causes harm / damage to body / cells  
*allow radiation sickness*  
 1  
 detail e.g., causes mutations / causes cancer / damages DNA /  
 damages chromosomes  
*allow two effects for 2 marks*  
 1
- [5]**

- 21** (a) (i) two protons 1
- 2 neutrons
- if neither point gained allow 1 mark for helium nucleus* 1
- (ii) electron 1
- (b) neutron splits (to form proton and electron) 1
- [4]

- 22** (i) 7 or 8 1
- correct data extracted from graph e.g. takes 8 days to drop from 50 to 25
- allow appropriate annotation of graph* 1
- (ii) long enough to destroy cancer cells
- do not accept dangerous unqualified* 1
- but short enough to minimise damage to surrounding tissues 1
- [4]

**23**

Quality of written communication: One mark for using correct scientific sequence :  
gravity → fusion → balance

1

any **four** from

- (dust and gas) pulled together by gravity
- (star formed when) it is hot enough  
*accept (as mass is pulled together) it gets very hot*
- hydrogen (and helium) nuclei fuse
- (these nuclear fusion reactions) release the energy / heat / light  
(which is radiated by stars)
- energy causes expansion
- gravitational pull is balanced by the expansion (force)

4

**[5]****24**

(a) (i) two protons

1

2 neutrons

*if neither point gained allow 1 mark for helium nucleus*

1

(ii) electron

1

(b) neutron splits (to form proton and electron)

1

**[4]****25**

Quality of written communication

correct use of **three** scientific terms e.g. radiation /  $\alpha$  or  $\beta$  or  $\gamma$  / cells /  
ionisation / mutation (not cells or body) / chromosomes / DNA /  
genes / cancer

1

any **three** from:

(materials emit) radiation

named type of radiation ( $\alpha$  or  $\beta$  or  $\gamma$ )

damage / harm / kill

*dangerous is neutral*

cells / chromosomes / DNA / genes

cancer

mutations

ionisation

gloves or glass absorb radiation / prevent radiation reaching body or cells

3

[4]

**26**

(a) protons

1

protons

*accept electrons*

1

neutrons

1

(b) protons

*reject mass*

1

[4]

**27**

2 weeks

*if answer is incorrect 2 gains two marks weeks gains one mark  
half of 68 or 34 gains one mark / allow working shown on graph*

[3]

**28**

- (i) (strontium-90)  
beta rays partly absorbed by aluminium

*accept gamma rays not absorbed **and** alpha all absorbed  
if phosphorus -32 then one mark max for beta ray explanation*

1

long half life means it can be used for a long time

1

- (ii) (technetium-99)  
(gamma) rays will pass out of body / less likely to be absorbed

*accept (gamma) less damaging or alpha / beta will damage cells if  
cobalt -60 then one mark max for gamma ray explanation*

1

short half life means it will not affect body over a long period

1

**[4]****29**

- neutron becomes proton / neutron emits electron / neutron emits beta particle

*gains proton neutral*

**[1]****30**

- (a) materials produced when earlier stars  
exploded

*accept the Sun is a second generation star  
accept formed from nebulae*

1



(b) **Quality of written communication:**

1 mark for correct sequencing balanced forces → expansion → contraction / explosion

1

any **five** from

gravity pulling matter together

*accept idea that a star is very massive so its force of gravity is very strong*

high temperatures that create expansion forces

*nuclear fusion releases energy that causes the very high temperatures*

these forces balance

star expands greatly

since expansion is greater than gravity

*accept fuel runs out*

forms a red giant

*give no further marks if red giant → white dwarf, red dwarf etc*

collapses inwards and explodes outwards

called a supernova

neutron star may form

leaves a small, dense object (a black hole)

*accept nothing can escape from it*

5

**[7]****31**

## (a) sensible scales

*full use of y axis*

1

completely accurate plotting

1

a smooth curve going through all but one of the points

*do not accept a dot-to-dot graph if two parts shown for curves  
accept the more correct*

1

at least one line or a clear mark showing how to obtain the half life  
from the graph and obtaining between 13 and 15

*at the bottom of the page cross or ticks in the order of the mark  
scheme*

1

- (b) (i) to let the beta particles get through  
*accept must be there to let the radiation through or if thick they may  
be stopped*

1

- (ii) alpha particles would be stopped by the glass **or** cannot penetrate glass  
*do not accept alphas are weak*

1

- (c) (i) it will give more counts per minute for a small quantity **or** it does not last  
so long so may not be as dangerous  
*accept answers in terms of 5 years assume it refers appropriately*

1

- (ii) it will not be there long enough to act as a tracer **or** it could cause  
radiation damage as all its activity will be in the first place it enters the system  
*accept answer in terms of 5 seconds  
accept not there long enough to work assume it refers appropriately*

1

[8]

32

any **one** of

- \* between (stage) 2 and (stage) 3
- \* (in) the main sequence
- \* (in) the main stable period
- \* (it is a) yellow dwarf

[1]

33

- (a) (i) alpha particles cannot penetrate covering  
*do not credit any answer not relating to film badge or its case*

1

- (ii) film gets fogged **or** blackened  
*accept film gets exposed  
do not credit film changes colour or goes white or blotchy*

1

- (b) (i) any **one** from  
 may cause cancer may damage cells **or** cell nuclei causes mutations  
 changes DNA  
*accept (causes) burns **or** kills cells*  
 1
- (ii) any **two** from  
 treating cancers  
 tracers in body  
 sterilising instruments **or** bandages  
*accept two descriptions of named treatments, eg thyroid check and  
 circulation monitoring*  
*accept is a source of X-rays, eg for dentistry **or** taking X-rays of  
 bones*  
 2
- (c) calculation that 1000 is 3 half lives on  
 $8000 \rightarrow 4000 \rightarrow 2000 \rightarrow 1000$   
 1
- time elapsed is  $3 \times \text{half life} = 31.8 \text{ hr}$   
**award both marks for 31.8 hr or 1 day 7.8 hr with no working shown**  
 1

[7]

34

- (a) (i) and (ii) in any order  
 1
- (i) alpha  
*accept Greek symbol ( $\alpha$ )*  
 1
- $\text{He}^{2+}$  **or**  ${}^4_2\text{He}$   
 1
- (ii) beta  
*accept Greek symbol ( $\beta$ ) **or** electron*  
 1
- $e^-$  **or**  ${}^0_{-1}e$   
*mass and atomic numbers are not required*  
*accept e*  
 1

- (b) (i) alpha  
*accept symbol* 1
- (ii) decreases  
then stops (entirely) **or** after a few cm  
*accept stops because  $\alpha$  can only travel a few cm in air* 1
- (c) it's gamma  
*accept its not ionising **or** it is not charged **or** it's not  $\alpha$  or  $\beta$  because a spark counter only measures  $\alpha$  or  $\beta$*  1
- [8]**

35

- (a) at least **6** points correctly plotted  
*gains 1 mark*
- (to better than half a square) **but all** points correctly plotted*  
*gains 2 marks* 2
- any **line** graph related to plotted points;  
point (3,29) discounted;  
best fit smooth curve  
*each for 1 mark* 3
- (b) radiation decreases with time  
*gains 1 mark*
- but** decreases quickly at first then more slowly  
*gains 2 marks*
- but** *idea that* it (about) halves every 2 weeks **or** half-life is (about) 2 weeks  
*gains 3 marks* 3
- [8]**

**36**

- (a) *evidence of conclusion*  $4 \times 1.007825$  **or** 4.0313  
*each gain 1 mark*

*based on use of data that there is a (very small) loss of mass*  
**or** 0.0276 **but** a loss of mass of 0.0276 **for every helium atom or** 0.69%/0.7%  
*gains 3 marks*

3

- (b) *idea that loss of mass results in release of energy*  
*gains 1 mark*

**but** small loss of mass results in huge energy release  
*gains 2 marks*

2

**[5]****37**

- (a) one relevant point correctly plotted  
*gains 1 mark*

**but** two relevant points correctly plotted  
*gains 2 marks*

**but** three relevant points correctly plotted  
*gains 3 marks*

curved line drawn accurately through the points  
*for 1 further mark*

4

- (b) age of igneous rock =  $400 \pm 100$  million years

1

- (c) sandstone is a sedimentary rock  
*for 1 mark*

there is likely to be some lead-207 present  
**or** from the rocks from which the sandstone was formed  
*for 1 mark*

(*allow*  $^{207}\text{Pb}$  may not have come from this  $^{235}\text{U}$ )

2

**[7]**

38

- (a) it use
- $E = mc^2$

$$\text{mass in kg i.e. } 0.001 \times \frac{0.7}{100}$$

*each gains 1 mark***but** 000007*gains 2 marks*

$$2.1 \times 10^3$$

*gains 3 marks*

evidence of 0.000007

mass in kg (i.e. 0.0007 **or** 0.7/100000)*each gains 1 mark*

squaring the speed of light

**but**  $6.3 \times 10^{11}$  (*credit alternative ways of stating this*)*gains 3 marks*

units J/joule

*for 1 further mark*

(N.B credit kJ, MJ, GJ but check power of 10 for full credit)

4

- (b) (i)
- idea that*
- the bigger the mass the shorter the life
- 
- gains 1 mark*

**but** *idea that* decrease in life is much more than  
proportional to increase in mass**or** more than proportional to mass<sup>2</sup>*gains 2 marks*

2

- (ii)
- ideas that:*
- 
- greater mass means greater
- core**
- temperature/pressure
- 
- greater core temperature/pressure means greater rate of fusion
- 
- increase in mass produces a proportionally much greater
- 
- increase in the rate of fusion

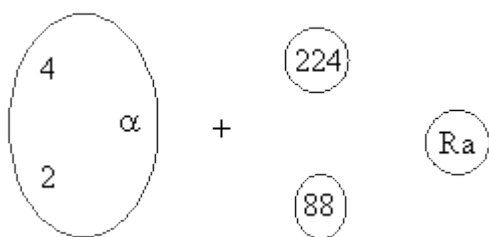
*each for 1 mark*

3

[9]

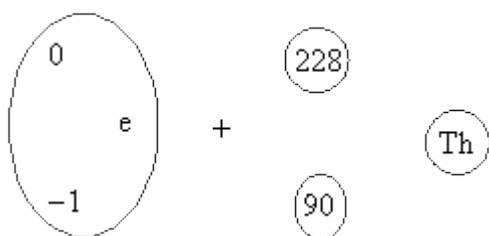
39

(a)

[Accept  $\text{He}^{2+}$  for  $\alpha$ ]each  for 1 mark

4

(b)

[Accept  $\beta$  for  $e$ ]each  for 1 mark

4

- (c) (i)    beta/  $\beta$             alpha/  $\alpha$   
           alpha/  $\alpha$             beta/  $\beta$   
           beta/  $\beta$  but    alpha/  $\alpha$   
           alpha/  $\alpha$             beta/  $\beta$   
           [i.e. consistent for 1; consistent and correct for 2]  
           *gains 2 marks*

2

(ii) *ideas that*

- many thorium atoms because they take so long to decay\*
- (many lead atoms because) the thorium has been decaying for so long/for billions of years
- or** (because) the rock is so/very/billions of years of years old
- many lead atoms because this is the stable end product [of the decay series]
- few atoms of other isotopes because they decay so quickly\*

[\*N.B. credit answers in terms of half-life]

*any three for 1 mark each*

3

**[13]**

40

(a) evidence of  $\frac{7350}{15}$   
gains 1 mark

**but**

490  
gains 2 marks

**but**

4900  
gains 3 marks

units  $\text{cm}^3$   
for 1 further mark

4

(b) some of radioactive solution gets into cells/body organs  
some of radioactive solution gets into urine (in the kidney)  
the radioactive solution becomes less radioactive during the test  
variability in readings  
in any order for 1 mark each

3

(c) ideas that

- won't lose (too) much radioactivity during the test
- won't stay radioactive/harm cells for too long after test is over  
for 1 mark each

2

[9]

41

- 4
- 9

each for 1 mark

[2]



42

(a)  $\gamma$ /gamma

- because more penetrating
- so can reach/damage cells from outside body/through skin

**but** $\alpha$ /alpha

- does more damage/more likely to cause cancer
- can only do this if inside the body/cells  
*each • for 1 mark*  
*[credit same ideas expressed conversely]*

4

(b) • must emit alpha /  $\alpha$  radiation

- *idea that* half-life must be just long enough to kill cancer cells  
*each for 1 mark*  
*[do not credit simply short half-life]*  
*(allow 'must be liquid / in solution)*

2

(c) *evidence of repeated halving then* $n$   $\approx$  3.6*gains 1 mark***but**

answer in range  
22 – 25.2 days  
(ie >6 and up to 7 half lives)

*gains 2 marks*

2

**[8]**

**43**

ideas that

- formed from dust/gases
- pulled together by gravity
- massive so very large gravitational forces (pulling inwards)
- hydrogen → helium / fusion releases energy [not fission or just 'nuclear']
- high temperature creates high pressure (pushing outwards)
- long period when forces balance
- then expands → red giant / red star
- then contracts to (dense) white dwarf / white star

*[credit if massive enough / more massive than sun, red giant → supernova → (very dense) neutron star but do not accept w.r.t. Sun itself]*

*[The whole of the (non bracketed part of) each idea must be present in some appropriate form of words for each mark to be credited. To gain more than a single mark ideas must also be in correct sequence and/or appropriately related.]*

*any six 1 mark each*

**[6]****44**

- A  $\beta$  / beta
- B  $\gamma$  / gamma
- C  $\alpha$  / alpha

*for 1 mark each*

**[3]**

45

- (a) indication (in writing or on graph) of finding point where radiation is halved (e.g. to 24 [from an initial 48]) and relating to the time difference between the two points

*gains 1 mark*

**but**

4.2-4.8\*

(\*i.e. in this range, including extremes)

*gains 2 marks*

units billions of years

*for 1 mark*

3

- (b)  $\frac{3}{4}$  **or** 75%  
[allow ecf from (a)]

*for 1 mark*

1

- (c) (i) *idea that* the intermediate nuclides are relatively short-lived

*for 1 mark*

1

- (ii) *idea that*  $\frac{1}{4}$  has decayed **or**  $\frac{3}{4}$  remains

*gains 1 mark*

**but**

read graph for radiation level of 36 (stated or shown on graph itself)

*gains 2 marks*

**but**

1.6-1.8\* (billion years)

(\* i.e. in this range, including extremes)

*gains 3 marks*

3

[8]

**46**

- (i) (fast moving) electrons (from the nucleus)  
(allow negatively charged particles)

*for 1 mark*

1

- (ii) protactinium has one neutron fewer  
protactinium has one proton more  
(credit has different numbers of neutrons / protons *with one mark*)

*for 1 mark each*

2

**[3]**